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Unit A2 2: Analytical, Transition Metals, Electrochemistry and Organic Nitrogen Chemistry 5.5 Transition metals 5.5.9 demonstrate understanding of the ligand replacement reactions of hexaaquacopper(II) ions with concentrated hydrochloric acid and ammonia solution, including colours and shapes of the complexes;

An equilibrium using copper(II) and ammonia | Experiment ...

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(Original post by trance addict) i think youre getting confused $q = mct$ $m =$ mass of liquid (so this would be in grams) $t =$ change in temperature C is a specific constant to the liquid youre calculating energy for, e.g. for water itd be $4.2 \text{ J g}^{-1} \text{ K}^{-1}$, you just put the number into the formula in the exam.. infact you can do that for all 3 quantities, you only need to specify the unit for the ...

$q = mct$ - The Student Room

Unit A2 2: Analytical, Transition Metals, Electrochemistry and Organic Nirtrogen Chemistry 5.2 Nuclear magnetic resonance chemistry 5.2.7 deduce a molecular structure from an nmr spectrum, limited to simple splitting patterns;

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